

BIOCHEMISTRY

Lec:3 : 2nd stage

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METABOLISM OF LIPIDS

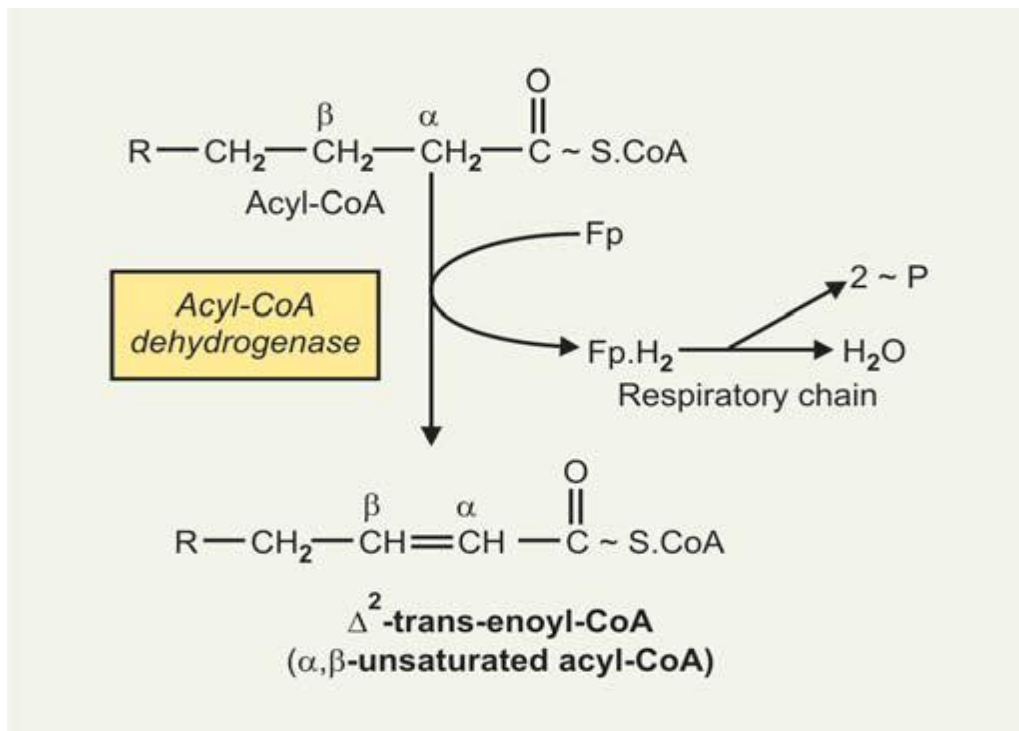
STEPS OF β -OXIDATION

Once acyl-CoA is transported by carnitine in the mitochondrial matrix, it undergoes β -oxidation by Fatty acid oxidase complex. The successive steps are as follows:

1. Dehydrogenation: Removal of 2 H atoms:

Removal of two hydrogen atoms from the 2 (α) and 3 (β) carbon atoms is catalysed by the enzyme **acyl-CoA dehydrogenase**, resulting in formation of Δ^2 -transenoyl-CoA (also called α , β -unsaturated acyl-CoA).

Hydrogen acceptor, i.e the co-enzyme for this dehydrogenase is a **flavo-protein**, containing FAD as prosthetic group, whose re-oxidation in the respiratory chain requires the mediation of another flavoprotein, called **electron transferring flavoprotein (ETF) + 2 ATP**



Types of acyl-CoA dehydrogenases: At least three acyl-CoA dehydrogenases have been described:

1. **“G”-green-coloured**, cu-containing, catalyses oxidation of fatty acids having chain length–C4 to C8.
2. **‘Y’ or ‘y1’-Yellow** flavo protein, catalyses oxidation of FA having chain length –C4 to C18, more specific for C6 **“hexonyl-CoA dehydrogenase”**
3. **‘y’ or ‘Y2’**-more active on FA having chain-length C6 to C18. Maximum activity on FA having chain-length C16 **“hexa-decanoyl dehydrogenase”**.

2. Hydration: Addition of one molecule of H₂O:

One molecule of water is added to saturate the double bond to form 3-OH acyl-CoA (called also as β -OH acyl-CoA), the reaction is catalysed by the enzyme **“ Δ 2 - enoyl -CoA hydratase”** (also called as Enoyl hydrolase).

3. Dehydrogenation: Removal of 2 hydrogen atoms:

The 3 – OH – Acyl-CoA undergoes further dehydrogenation on the 3 carbon, catalysed by the enzyme **3–OH–acyl-CoA dehydrogenase**, to form the corresponding 3–ketoacyl-CoA (β -ketoacyl-CoA).

Hydrogen acceptor, i.e. coenzyme of this dehydrogenase is NAD⁺. Reduced NAD when oxidized in respiratory chain produces 3 ATP.

4. Thiolytic cleavage:

Finally, 3-keto-acyl-CoA is split at the 2,3 position by **thiolase** (**“3 – keto acyl thiolase”** or **“acetyl–CoA acyl transferase”**), which catalyses a thiolytic cleavage involving another molecule of CoA.

End-products of this reaction: The thiolytic cleavage results in formation of:

1. One molecule of acetyl-CoA .
2. An acyl-CoA molecule containing 2-carbons less than the original acyl-CoA molecule, which enters for oxidation by the enzyme acyl-CoA dehydrogenase (reenters at step 1).

In this way, a long-chain FA may be degraded completely to “acetyl-CoA” (C-2 units). Acetyl-CoA can be oxidized to CO₂ and H₂O and thus complete oxidation of FA is achieved. Thus, end-product of β -oxidation of a

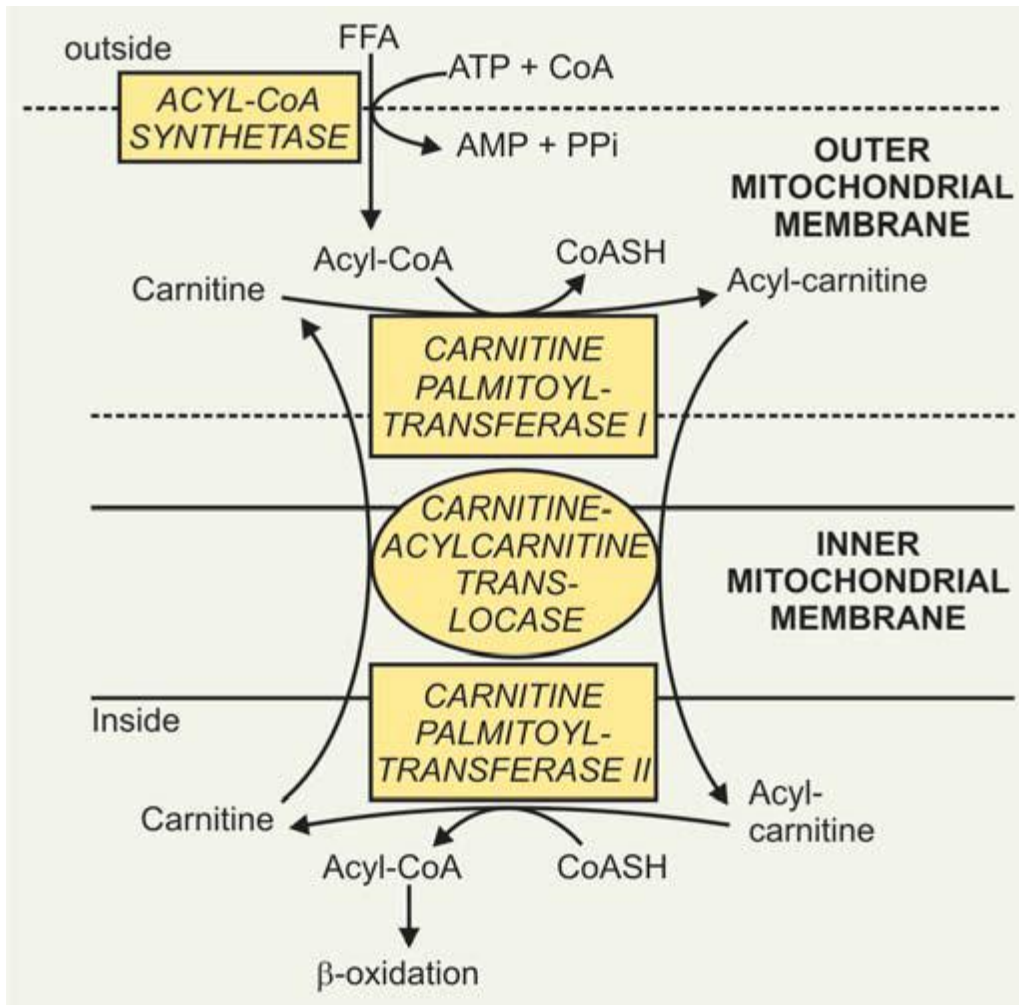
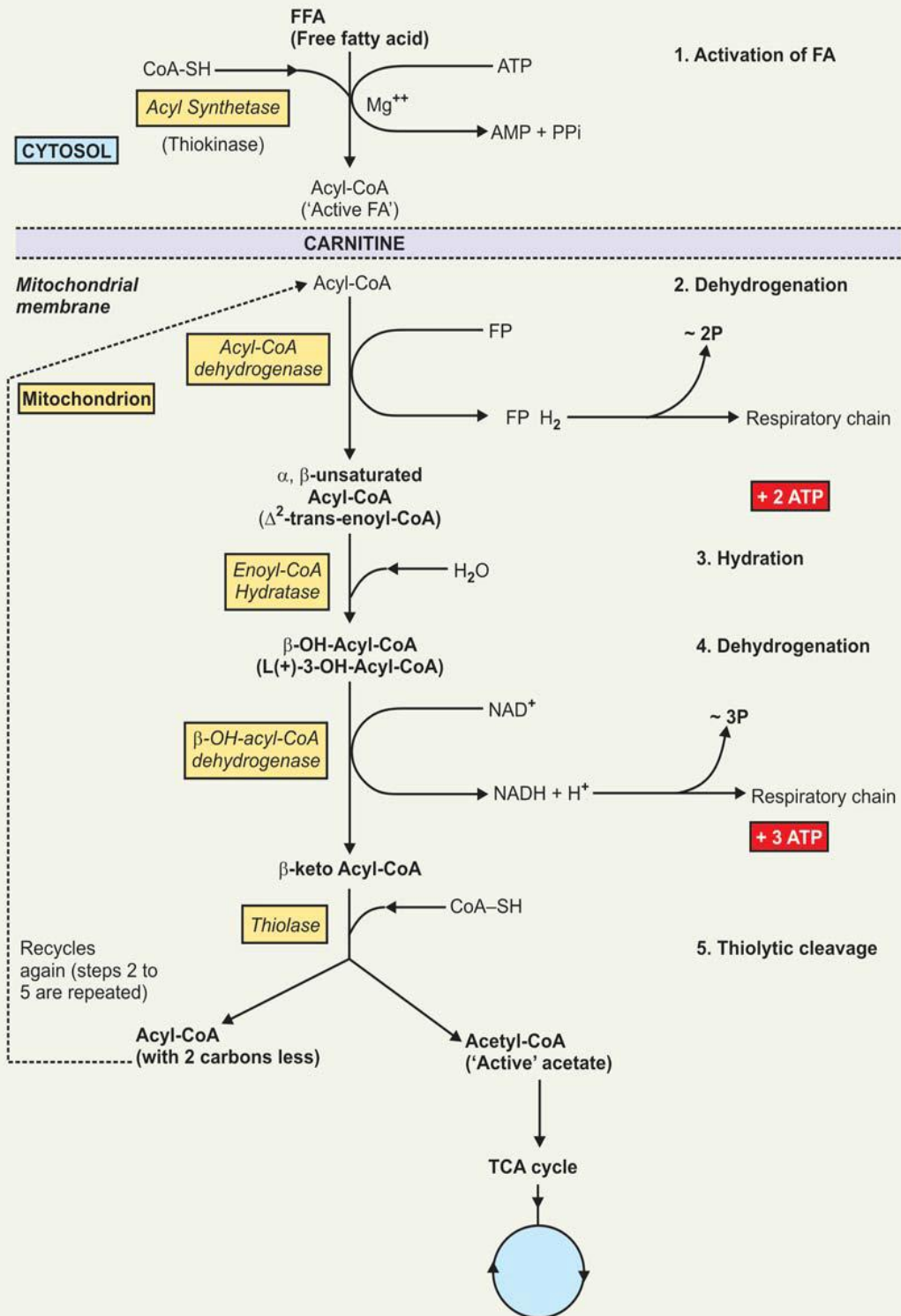
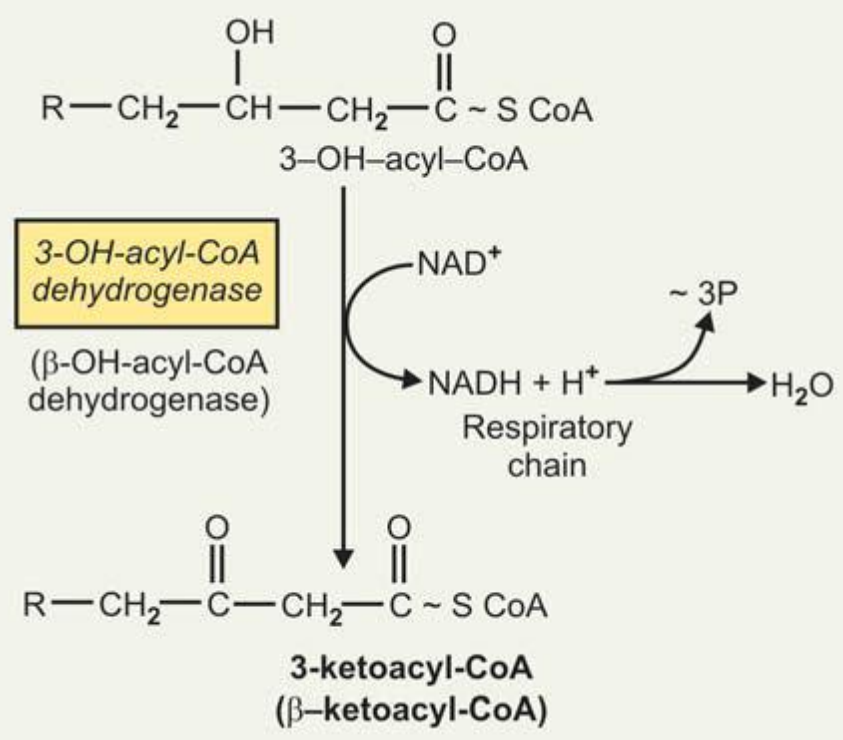
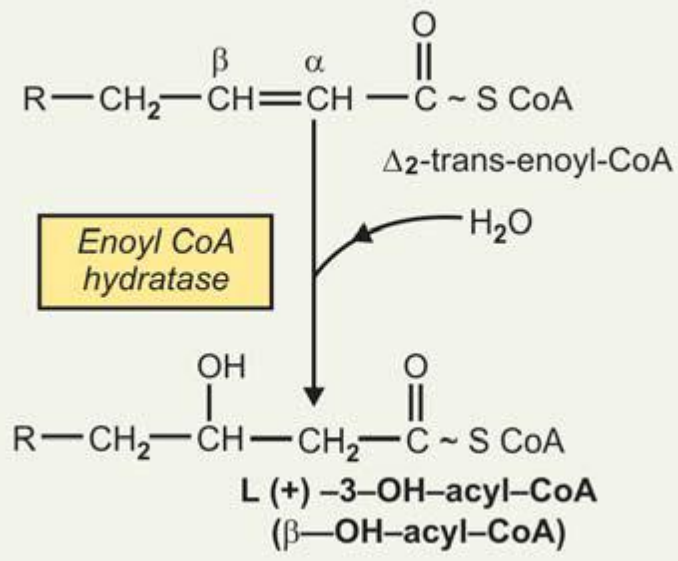
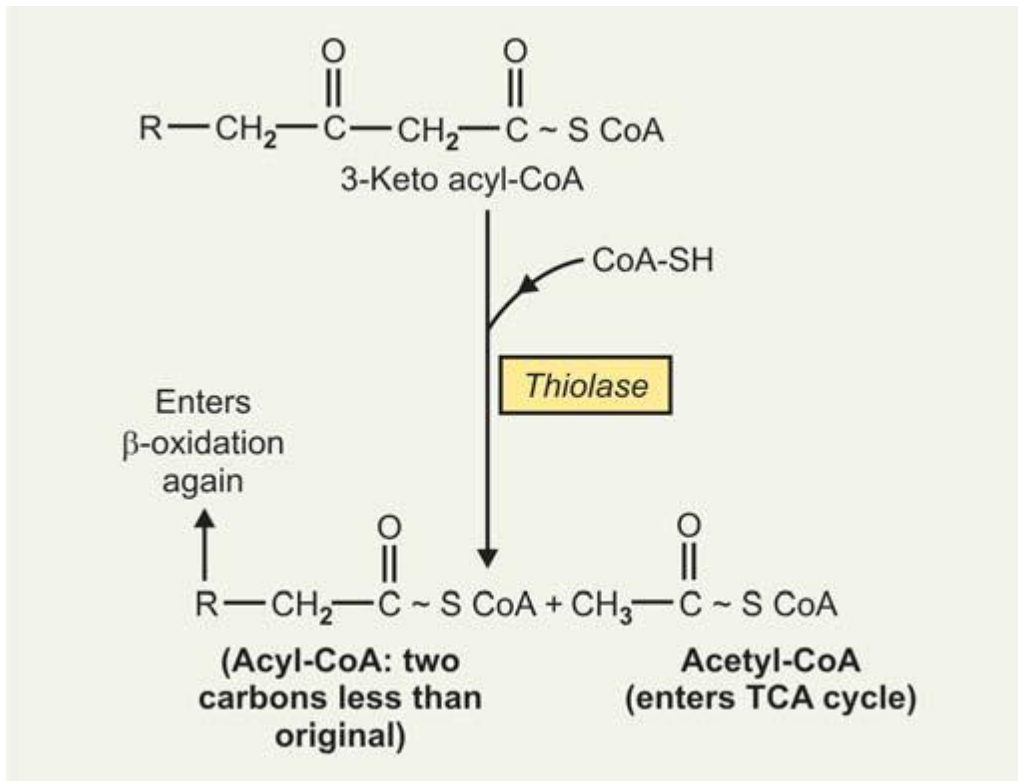


Fig: Role of carnitine in the transport of long-chain FA







How many acetyl-CoA are produced from β -oxidation of palmitic acid?

Palmitic acid is C₁₆ H₃₂ O₂. In β -oxidation, it will **undergo 7 (seven) cycles**, producing 7 acetyl-CoA (in 7 cycles) + 1 acetyl-CoA (last cycle-one extra).

\therefore Total acetyl-CoA produced by β -oxidation of one molecule of palmitic acid = 8 acetyl-CoA.

β -Oxidation of FA with an odd number of carbon Atoms: Fatty acids with an odd number of carbon atoms are oxidized by β -oxidation pathway to produce acetyl-CoA until a 3-carbon residue **propionyl-CoA** is left.

Propionyl-CoA is metabolised to succinyl-CoA through methyl malonyl-CoA.

Bioenergetics of β -Oxidation and its Efficiency

Palmitic acid, $C_{15}H_{31}COOH$, on complete oxidation (β -oxidation) **produces 8 acetyl-CoA** (Refer discussion above). Transport of electrons in respiratory chain from reduced Fp and NAD in each cycle produces 5 (five) high energy phosphate bonds.

Hence, 7 cycles $(7 \times 5) = 35 \sim P$

Total 8 molecules of acetyl-CoA,

When oxidised in TCA cycle will

produce $= 12 \times 8 = 96 \sim P$

Total high energy phosphate bonds

produced $= 131 \sim P$

Total = 131 $\sim P$

In initial activation of

FA $\sim P$ bond utilised $= -2 \sim P$

\therefore Total gain = 129 $\sim P$

\therefore Energy Production $= 129 \times 7.6 = 980 \text{ Kc}$

(or $129 \times 30.5 = 3935 \text{ Kj}$)

Caloric value of Palmitic acid

(Bomb calorimeter) $= 2340 \text{ Kc/mol}$

Hence, **efficiency $= 980/2340 \times 100 = 41\%$ of the total energy of combustion of FA.**

B. α -OXIDATION

α -oxidation is another alternative pathway for oxidation of FA which involves decarboxylation of the COOH group after hydroxylation and the formation of a FA containing an “odd” number of carbon atoms, which subsequently undergoes repeated β -oxidation. **No initial activation of FA is necessary in this process.**

C. ω -OXIDATION (VERKADE)

In ω -oxidation, Fatty acids undergo oxidation at the carbon atom farthest removed from the carboxyl group (ω -carbon) producing a *dicarboxylic acid*, which is then subjected to β -oxidation and cleavage to form successively smaller dicarboxylic acids. Both processes occur principally in brain microsomes but are negligible in extent as compared to β -oxidation.

Essential differences and similarities between α -oxidation and ω -oxidation are shown in [Table](#).

Table 1: Differences in α -and ω -oxidation

α -oxidation	ω -oxidation
1. Substrate: Even carbon long-chain FA (some of them)	1. Some medium- and long-chain FA
2. Presence of O₂ —oxidised aerobically in presence of O ₂	2. Occurs aerobically in presence of O ₂
3. Sites: Microsomes of brain and liver	3. Liver microsomes
4. Enzyme: α -hydroxylase—a monooxygenase	4. FA ω -hydroxylase also a monooxygenase
5. Cofactors required —Fe ⁺⁺ , Vitamin C/FH ₄	5. Cytochrome P450, flavoprotein reductase and NADP ⁺
6. No initial activation of FA is required	6. No initial activation of FA is necessary.
7. Steps: (i) Step-1: Formation of α -OH-FA $\begin{array}{c} \text{R} - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{COOH} \\ \downarrow \text{Fe}^{++} \\ \alpha\text{-hydroxylase} \quad \text{Vit C/or FH}_4 \\ \text{OH} \\ \\ \text{R} - \text{CH}_2 - \text{CH}_2 - \text{CH} - \text{COOH} \\ \alpha\text{-OH-FA} \end{array}$	7. Steps: (i) Step-1: Hydroxylation of ω carbon. $\begin{array}{c} \text{CH}_3 - \text{CH}_2 - (\text{CH}_2)_n - \text{CH}_2 - \text{COOH} \\ \downarrow \text{O}_2 \\ \omega\text{-hydroxylase} \\ \text{H}_2\text{O} \leftarrow \text{Cyt P 450} \\ \text{OH} \\ \\ \text{CH}_2 - \text{CH}_2 - (\text{CH}_2)_n - \text{CH}_2 - \text{COOH} \\ \omega\text{-OH-FA} \end{array}$ <p style="text-align: right;">Even C – FA</p>

<p>(ii) Step-2: Decarboxylation to produce an</p> $ \begin{array}{c} \text{OH} \\ \\ \text{R} - \text{CH}_2 - \text{CH}_2 - \text{CH} - \text{COOH} \\ \downarrow \text{Dehydrogenase} \quad \text{NAD}^+ \\ \text{NADH} + \text{H}^+ \\ \text{O} \\ \\ \text{R} - \text{CH}_2 - \text{CH}_2 - \text{C} - \text{COOH} \\ \alpha\text{-keto acid} \\ \swarrow \searrow \\ \text{CO}_2 \\ \text{R} - \text{CH}_2 - \text{CH}_2 - \text{COOH} \\ \text{Odd chain FA with one carbon less} \end{array} $	<p>(ii) Step-2: ω-OH FA is oxidised with the help of an NADPdependent enzyme to produce- α-ω dicarboxylic acid</p> $ \begin{array}{c} \text{OH} \\ \\ \text{CH}_2 - \text{CH}_2 - (\text{CH}_2)_n - \text{CH}_2 - \text{COOH} \\ \omega\text{-OH FA} \\ \downarrow \text{Enz} \quad \text{NADP}^+ \\ \text{H} \\ \\ \text{C} - \text{CH}_2 - (\text{CH}_2)_n - \text{CH}_2 - \text{COOH} \\ \\ \text{O Aldehyde} \\ \downarrow \text{Enz O}_2 \\ \text{HOOC} - \text{CH}_2 - (\text{CH}_2)_n - \text{CH}_2 - \text{COOH} \\ \alpha, \omega\text{-dicarboxylic acid} \end{array} $
<p>(iii) Step-3: Odd chain FA undergoes repeated β-oxidation to produce (Acetyl-CoA)$_n$ + Propionyl-CoA"</p> $ \begin{array}{c} \text{Propionyl-CoA} \\ \downarrow \\ \text{Succinyl-CoA} \\ \downarrow \\ \text{TCA cycle} \end{array} $	<p>(iii) Step-3: Even carbon dicarboxylic acid then undergoes-repeated β-oxidation to yield (Acetyl-CoA)$_n$ and one molecule of succinyl-CoA (intermediate in TCA Cycle).</p> $ \begin{array}{c} \downarrow \\ \text{TCA cycle} \end{array} $